

page	place	part	present	correct
9	SYMBOLS AND UNITS	$c_p$	Isobaric heat capacity	isobaric heat capacity
9	SYMBOLS AND UNITS	$c_v$	Isochoric heat capacity	isochoric heat capacity
10	CONVERSION FACTOR	Pressure	1atm = 1.03125 bar	1atm = 1.01325 bar
10	CONVERSION FACTOR	Density	(Unit) $\text{ft}^3 \cdot \text{lbm}^{-1}$	(Unit) $\text{lbm} \cdot \text{ft}^{-3}$
10	CONVERSION FACTOR	Density	$1 \text{ g} \cdot \text{cm}^{-3} = 65.4280 \text{ ft}^3 \cdot \text{lbm}^{-1}$	$1 \text{ g} \cdot \text{cm}^{-3} = 62.4280 \text{ lbm} \cdot \text{ft}^{-3}$
10	CONVERSION FACTOR	Energy	$1 \text{ kcal} = 1.6222 \times 10^{-3} \text{ kWh}$	$1 \text{ kcal} = 1.16222 \times 10^{-3} \text{ kWh}$
10	CONVERSION FACTOR	Energy	$1 \text{ kcal}_{\text{IT}} = 1.6300 \times 10^{-3} \text{ kWh}$	$1 \text{ kcal}_{\text{IT}} = 1.16300 \times 10^{-3} \text{ kWh}$
10	CONVERSION FACTOR	Energy	$1 \text{ kcal} = 3.06567 \text{ Btu}_{\text{IT}}$	$1 \text{ kcal} = 3.96567 \text{ Btu}_{\text{IT}}$
11	CONVERSION FACTOR	Specific enthalpy	(Unit) $\text{Btu}_{\text{IT}} \cdot \text{kg}^{-1}$	(Unit) $\text{Btu}_{\text{IT}} \cdot \text{lbm}^{-1}$
11	CONVERSION FACTOR	Specific enthalpy	(Unit) $1 \text{ Btu}_{\text{IT}} \cdot \text{kg}^{-1} =$	(Unit) $1 \text{ Btu}_{\text{IT}} \cdot \text{lbm}^{-1} =$
11	CONVERSION FACTOR	Specific entropy	$1 \text{ kJ} \cdot \text{kg}^{-1} \cdot \text{K}^{-1} = 0.429923 \text{ Btu} \cdot \text{lbm}^{-1} \cdot ^\circ\text{R}^{-1}$	$1 \text{ kJ} \cdot \text{kg}^{-1} \cdot \text{K}^{-1} = 0.238846 \text{ Btu} \cdot \text{lbm}^{-1} \cdot ^\circ\text{R}^{-1}$
11	CONVERSION FACTOR	Specific entropy	$1 \text{ kcal} \cdot \text{kg}^{-1} \cdot \text{K}^{-1} = 1.79880 \text{ Btu} \cdot \text{lbm}^{-1} \cdot ^\circ\text{R}^{-1}$	$1 \text{ kJ} \cdot \text{kg}^{-1} \cdot \text{K}^{-1} = 0.99933 \text{ Btu} \cdot \text{lbm}^{-1} \cdot ^\circ\text{R}^{-1}$
11	CONVERSION FACTOR	Thermal conductivity	(Unit) $\text{Btu}_{\text{IT}} \cdot \text{ft}^{-1} \cdot \text{h}^{-1} \cdot ^\circ\text{R}^{-1}$	(Unit) $\text{Btu}_{\text{IT}} \cdot \text{ft}^{-1} \cdot \text{h}^{-1} \cdot ^\circ\text{F}^{-1}$
11	CONVERSION FACTOR	Thermal conductivity	(Unit) $1 \text{ Btu}_{\text{IT}} \cdot \text{ft}^{-1} \cdot \text{h}^{-1} \cdot ^\circ\text{R}^{-1} =$	(Unit) $1 \text{ Btu}_{\text{IT}} \cdot \text{ft}^{-1} \cdot \text{h}^{-1} \cdot ^\circ\text{F}^{-1} =$
12	Table S.1	R50	$M = 16.042 \text{ kg/kmol}$	$M = 16.043 \text{ kg/kmol}$
12	Table S.1	R134a	$v_c = 199 \text{ cm}^3/\text{mol}$	$v_c = 200 \text{ cm}^3/\text{mol}$
12	Table S.1	R143a	$v_c = 196 \text{ cm}^3/\text{mol}$	$v_c = 194 \text{ cm}^3/\text{mol}$
12	Table S.1	R152a	$v_c = 180 \text{ cm}^3/\text{mol}$	$v_c = 179 \text{ cm}^3/\text{mol}$
21	Table S.7	R225ca	$\text{CF}_2\text{CF}_2\text{CHCl}_2$	$\text{CF}_3\text{CF}_2\text{CHCl}_2$
33	(Title)		2. EQUATI <del>OS</del> OF STATE	2. EQUATION <del>S</del> OF STATE
43	line 9		... and $T_r^0 = 273.15 \text{ K}$ .	... and $T^0 = 273.15 \text{ K}$ .
45	Table 3-2		R-141b	R141b
45	Table 3-2		R-225ca	R225ca
45	Table 3-2		R-225cb	R225cb
112 to 131	Tables T.1 to T.10	$h'$	$\text{J} \cdot \text{kg}^{-1}$	$\text{kJ} \cdot \text{kg}^{-1}$
112 to 131	Tables T.1 to T.10	$h''$	$\text{J} \cdot \text{kg}^{-1}$	$\text{kJ} \cdot \text{kg}^{-1}$
112 to 131	Tables T.1 to T.10	$s'$	$\text{J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$	$\text{kJ} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$
112 to 131	Tables T.1 to T.10	$s''$	$\text{J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$	$\text{kJ} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$
132 to 135	Tables T.11 to T.12	$h''-h'$	$\text{J} \cdot \text{kg}^{-1}$	$\text{kJ} \cdot \text{kg}^{-1}$
132 to 135	Tables T.11 to T.12	$s''-s'$	$\text{J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$	$\text{kJ} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$